# ORGANIC CHEMISTRY LABORATORY TECHNIQUES

# Lisa Nichols, 2<sup>nd</sup> Edition









## TABLE OF CONTENTS, IN BRIEF

Preface	Chapter 1 General Techniques Pg. 15	Chapter 2 Chromatography Pg. 81	Chapter 3 Crystallization Pg. 155
<ul><li>Table of Contents</li><li>About the Author</li></ul>	<ul><li>Glassware, Clamping</li><li>Transferring Solids</li></ul>	<ul><li>Generalities</li><li>Thin Layer</li></ul>	• Single Solvent Crystallization
• Using this Resource	and Liquids	Chromatography	• Mixed Solvent Crystallization
	• Inert Atmospheric Methods	Column     Chromatography	Crystamzation
	• Heating	Gas Chromatography	
	• Refluxing		
	• Filtering		
Chapter 4 Extraction Pg. 201	Chapter 5 Distillation Pg. 247	Chapter 6 Miscellaneous Techniques Pg. 307	Chapter 7 Technique Summaries Pg. 361
			<image/> <image/> <image/> <image/> <image/> <table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row><table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row></table-row>
• Single Extraction	• Simple Distillation	• Melting Point	Technique Summaries
Multiple Extractions	Fractional Distillation	Boiling Point	
<ul><li> Reaction Work-ups</li><li> Drying Agents</li></ul>	<ul><li>Vacuum Distillation</li><li>Steam Distillation</li></ul>	<ul><li>Sublimation</li><li>Chemical Tests</li></ul>	
21,116/160103	Rotary Evaporation		

### Copyright © 2017 Lisa Nichols

#### Butte Community College

All photographs (unless otherwise indicated) were taken by Lisa Nichols.

The author has received written permission to use the photograph of all students appearing in this textbook.

## TABLE OF CONTENTS, IN DETAIL

Chapter 0: Preface	
About the Author	11
Using this Resource	12

### Chapter 1: General Techniques

1.1	Glassware and Equipment	17
1.1.A	Pictures of Glassware + Equipment	17
1.1.B	Ground Glass Joints	20
1.1.C	Clamping	21
1.1.D	Greasing Joints	23
1.1.E	Cleaning Glassware	24
1.1.F	Drying Glassware	24
	Quick Drying	24
	Oven and Flame Drying	25
	Drying Tubes	26
1.1.G	Storing Samples (Parafilm/Teflon Tape)	26
1.2	Transferring Methods	27
1.2 1.2.A	Solids	27
		27
1.2.B	Liquids	
	Pouring Liquids	28
	Comments Re: Measurements	28
	Using Pasteur Pipettes	29
	Using Calibrated Pipettes	30
	Calibrated Plastic Pipettes	30
	Calibrated Glass Pipettes	30
	Calibrated Pipettes Summary	34
	Dispensing Highly Volatile Liquids	35
	Pouring Hot Liquids	35
1.2.C	Inert Atmospheric Methods	36
	Step-by-Step Procedures	37
	Inert Atmospheric Methods Summary	43

1.3	Heating and Cooling Methods	44
1.3.A	Methods and Flammability	44
1.3.B	Controlled Boiling	46
	Boiling Stones (Boiling Chips)	46
	Boiling Sticks (Wood Splints)	47
	Stir Bars and Spin Vanes	47
1.3.C	Adjustable Platforms	48
1.3.D	Bunsen Burners	49
1.3.E	Hotplates	51
1.3.F	Steam Baths	52
1.3.G	Heating Mantles	53
1.3.H	Water, Sand, and Oil Baths	54
1.3.I	Heat Guns	56
1.3.J	Cooling Baths	57
1.3.K	Reflux	58
	Overview of Reflux	58
	Step-by-Step Procedures	58
	Reflux Summary	62
1.4	Reflux Summary Filtering Methods	62 63
<b>1.4</b> 1.4.A		-
	Filtering Methods	63
1.4.A	<b>Filtering Methods</b> Overview of Methods	<b>63</b> 63
1.4.A 1.4.B	<b>Filtering Methods</b> Overview of Methods Decanting	<b>63</b> 63 63
1.4.A 1.4.B 1.4.C	<b>Filtering Methods</b> Overview of Methods Decanting Gravity Filtration	<b>63</b> 63 63 64
1.4.A 1.4.B 1.4.C	<b>Filtering Methods</b> Overview of Methods Decanting Gravity Filtration Suction Filtration	<b>63</b> 63 63 64 65
1.4.A 1.4.B 1.4.C	<b>Filtering Methods</b> Overview of Methods Decanting Gravity Filtration Suction Filtration Suction Filtration Overview	<b>63</b> 63 63 64 65 65
1.4.A 1.4.B 1.4.C	Filtering Methods Overview of Methods Decanting Gravity Filtration Suction Filtration Suction Filtration Overview Rinsing	<b>63</b> 63 63 64 65 65 65
1.4.A 1.4.B 1.4.C	Filtering Methods Overview of Methods Decanting Gravity Filtration Suction Filtration Suction Filtration Overview Rinsing Water Aspirator	<b>63</b> 63 64 65 65 66 67
1.4.A 1.4.B 1.4.C	Filtering Methods Overview of Methods Decanting Gravity Filtration Suction Filtration Suction Filtration Overview Rinsing Water Aspirator Step-by-Step Procedures	<b>63</b> 63 63 64 65 65 66 67 68
1.4.A 1.4.B 1.4.C 1.4.D	Filtering Methods Overview of Methods Decanting Gravity Filtration Suction Filtration Suction Filtration Overview Rinsing Water Aspirator Step-by-Step Procedures Suction Filtration Summary	63 63 64 65 65 66 67 68 72
1.4.A 1.4.B 1.4.C 1.4.D	Filtering Methods Overview of Methods Decanting Gravity Filtration Suction Filtration Suction Filtration Overview Rinsing Water Aspirator Step-by-Step Procedures Suction Filtration Summary Hot Filtration	63 63 64 65 65 66 67 68 72 73
1.4.A 1.4.B 1.4.C 1.4.D	Filtering Methods Overview of Methods Decanting Gravity Filtration Suction Filtration Suction Filtration Overview Rinsing Water Aspirator Step-by-Step Procedures Suction Filtration Summary Hot Filtration Hot Filtration Overview	63 63 63 64 65 65 66 67 68 72 73 73
1.4.A 1.4.B 1.4.C 1.4.D	Filtering Methods Overview of Methods Decanting Gravity Filtration Suction Filtration Suction Filtration Overview Rinsing Water Aspirator Step-by-Step Procedures Suction Filtration Summary Hot Filtration Hot Filtration Overview Step-by-Step Procedures	63 63 63 64 65 65 66 67 68 72 73 73 73 75

# Chapter 2: Chromatography

81

2.1	Chromatography Generalities	83
2.1.A	Overview of Chromatography	84
2.1.B	General Separation Theory	85
2.2	Thin Layer Chromatography (TLC)	86
2.2.A	Overview of TLC	86
2.2.B	Uses of TLC	87
	Assessing Purity	87
	Assessing Reaction Progress	88
	Use of a Co-spot	88
	Monitoring a Reaction by TLC	89
	Obtaining an Aliquot	90
2.2.C	The Retention Factor $(R_f)$	91
2.2.D	Separation Theory	93
	General Theory	93
	Structural Considerations	95
	Mobile Phase Polarity	97
2.2.E	Step-by-Step Procedures	98
	General TLC Procedure	98
	Thin Layer Chromatography Summary	102
	Troubleshooting	103
	Making Capillary TLC Spotters	104
	Notebook Record of TLC's	105
2.2.F	Visualizing TLC Plates	106
	Visualization Summary	106
	Ultraviolet Absorption	107
	Iodine	108
	Chemical Stains	109
	General Staining Procedure	109
	<i>p</i> -Anisaldehyde / Vanillin Stains	110
	Permanganate Stain	112
	Phosphomolybdic Acid Stain	113
	Iron(III) Chloride Stain	113
	Bromocresol Green Stain	114
	Visualization Troubleshooting	115
	6	

2.3	Column Chromatography	119
2.3.A	Macroscale Columns	120
	Procedural Generalities	120
	Step-by-Step Procedures	121
	Macroscale Column Summary	131
	Troubleshooting	132
2.3.B	Microscale (Pipette) Columns	133
	Step-by-Step Procedures	133
	Microscale Column Summary	138
2.4	Gas Chromatography (GC)	139
2.4.A	Overview of GC	139
2.4.B	Uses of GC	140
	Assessing Purity	140
	Identifying Components	141
2.4.C	Separation Theory	142
	General Theory	142
	Structural Considerations	144
2.4.D	Quantitating with GC	145
	Quantitative Challenges	145
	Using a Calibration Curve	147
	Using Response Factors	148
2.4.E	GC Parameters	149
	Dilution	149
	Solvent Delay	150
	Oven Temperature	151
	Using a Temperature Ramp	152
2.4.F	Sample Preparation	153

# Chapter 3: Crystallization

3.1	<b>Overview of Crystallization</b>	157
3.2	Uses of Crystallization	158
3.3	Choice of Solvent	160
3.3.A	Ideal Temperature Profile	160
3.3.B	General Procedures for Removing Impurities	161
3.3.C	Determining Which Solvent to Use	162
3.3.D	Using Solubility Data	163
3.3.E	Experimentally Testing Solvents	164
3.3.F	Mixed Solvents	165
3.4	<b>Crystallization Theory</b>	167
3.4.A	Purification	167
3.4.B	Cooling Slowly	167
3.4.C	Using the Minimum Amount of Hot Solvent	169
3.4.D	The Unavoidable Loss of Recovery	171
3.4.E	Quantitating Crystallization	173
	With a Soluble Impurity	173
	With an Impurity of Similar Solubility	175
3.4.F	Second Crop Crystallization	176

3.5	<b>Procedural Generalities</b>	179
3.5.A	General Procedure	179
3.5.B	Heat Source and Bump Prevention	180
3.5.C	Charcoal	181
3.5.D	Cooling Slowly	183
3.5.E	Initiating Crystallization	184
3.6	Step-by-Step Procedures	186
3.6.A	Single Solvent Crystallization	186
3.6.B	Crystallization Summary	190
3.6.C	Using Solvents Other Than Water	191
	Ethanol, Methanol, Ethyl Acetate, and Hexanes	191
	Diethyl Ether, Acetone, and Petroleum Ether (Low-Boiling)	192
3.6.D	Mixed Solvent Crystallization	193
3.6.E	Mixed Solvent Summary	195
3.6.F	Troubleshooting	196
	Crystallization is too Quick	196
	Crystallization Doesn't Happen	197
	The Yield is Poor	198
	Liquid Droplets Form (The Solid Oils Out)	199

# Chapter 4: Extraction 201

4.1	<b>Overview of Extraction</b>	203
4.2	Uses of Extraction	204
4.2.A	Extracting Natural Compounds	204
4.2.B	Transferring Compounds from Layers	204
4.2.C	Selective Removal of Components	205
4.3	Which Layer is Which?	206
4.3.A	Density	206
4.3.B	How To Determine the Aqueous Layer	207
4.4	Extraction Theory	208
4.4.A	Partition / Distribution Coefficient (K)	208
4.4.B	Choosing a Solvent with Solubility Data	209
4.4.C	Quantitating Single Extraction	210
4.4.D	Multiple Extractions	211
	Overview of Multiple Extraction	211
	Quantitating Multiple Extraction	213

4.5	Step-by-Step Procedures	215
4.5.A	Single Extraction	215
4.5.B	Single Extraction Summary	221
4.5.C	Multiple Extractions	222
	Organic Layer is on the Top	222
	Organic Layer is on the Bottom	224
4.5.D	Troubleshooting	225
	There is Only One Layer	225
	There are Three Layers	225
	There in Insoluble Material at the Interface	225
	The Interface Cannot Be Seen	226
	The Layers Don't Separate Well (An Emulsion Formed)	226
4.5.E	Microscale Extractions	228
4.6	<b>Reaction Work-ups</b>	230
4.6.A	Purpose of a Work-up	230
4.6.B	Common Washes	230
	Water	230
	Sodium Bicarbonate and Sodium Carbonate	232
	Brine (Saturated NaCl)	234
4.6.C	Drying Agents	237
	Why They are Used	237
	Types of Drying Agents	238
	Drying Agents Procedure	240
4.7	Acid-Base Extraction	242
4.7.A	How They Work	242
4.7.B	Sodium Bicarbonate Washes	243
4.7.C	Mixtures of Acids and Bases	244
	Extracting Bases	244
	Extracting Carboxylic Acids Vs. Phenols	244
	Extracting Acid, Base, and Neutral Components	245

### Chapter 5: Distillation 247

5.1	<b>Overview of Distillation</b>	249
5.2	Simple Distillation	251
5.2.A	Uses of Simple Distillation	251
	Concentration of Alcohol	251
	Distilled Water	252
	Purification of Reagents + Products	252
5.2.B	Separation Theory	254
	Raoult's and Dalton's Laws	254
	Purification Potential	255
	Distillation Curves	258
	Distilling Temperatures	260
	Azeotropes	262
5.2.C	Step-by-Step Procedures	264
	Condenser Hoses	264
	Simple Distillation Procedure	265
	Simple Distillation Summary	272
	Variations	273
	Troubleshooting	274
	Insulation	274
5.2.D	Microscale Distillation	276
	Semi-Microscale	276
	Hickman Head	276
	Short-Path Distillation	277
5.3	Fractional Distillation	278
5.3.A	Theory of Fractional Distillation	278
5.3.B	Fractionating Columns	279
5.3.C	Uses of Fractional Distillation	281
	Oil Refining	281
	Purification of Reagents + Products	281
5.3.D	Step-by-Step Procedures	282

Fractional Distillation Procedure

Fractional Distillation Summary

282

284

5.4	Vacuum Distillation	285
5.4.A	Overview of Vacuum Distillation	285
5.4.B	Predicting the Boiling Temperature	286
5.3.C	Step-by-Step Procedures	287
	Vacuum Distillation Procedure	287
	Vacuum Distillation Summary	291
5.5	Steam Distillation	292
5.5.A	Overview of Steam Distillation	292
5.5.B	Uses of Steam Distillation	292
5.5.C	Separation Theory	293
5.5.D	Step-by-Step Procedures	294
	Steam Distillation Procedure	294
	Steam Distillation Summary	299
5.6	<b>Rotary Evaporation</b>	300
5.6.A	Overview of Rotary Evaporation	300
5.6.B	Step-by-Step Procedures	301
	Rotary Evaporation Procedure	301
	Rotary Evaporation Summary	304
5.6.C	Troubleshooting	305

## Chapter 6: Miscellaneous Techniques 307

6.1	Melting Point	309	
6.1.A	Overview of Melting Point	309	
6.1.B	Uses of Melting Point	310	
	Identification	310	
	Assessing Purity	311	
6.1.C	Melting Point Theory	313	
	Melting Point Diagrams	313	
	Impurities Effect on the Melting Point	314	
	Melting Point Depression	314	
	Broadening of the Melting Point	315	
6.1.D	Step-by-Step Procedures	316	
	Sample Preparation	316	
	Melting Point Apparatus	317	
	Thiele Tube Method	320	
	Melting Point Summary	322	
6.1.E	Mixed Melting Points	323	
6.2	Boiling Point	324	
6.2.A	Overview of Boiling Point	324	
6.2.B	Step-by-Step Procedures	324	
	Distillation Method	324	
	Reflux Method	325	
	Thiele Tube Method	326	
	Thiele Tube Theory	326	
	Thiele Tube Procedure	327	
	Thiele Tube Summary	329	
Chapter 7: Technique Summaries 361			
7.1	Flame-Drying Glassware	362	
7.2	Using Calibrated Glass Pipettes	363	
7.3	Inert Atmospheric Methods	364	
7.4	Reflux	365	
7.5	Suction Filtration	366	
7.6	Hot Filtration	367	
7.7	Thin Layer Chromatography	368	
7.8	TLC Visualization Methods	369	
7.9	Macroscale Column Chromatography	370	
7.10	Pipette Column Chromatography	371	
7.11	Testing Solvents for Crystallization	372	
7.12	Testing Mixed Solvents for Crystallization	373	
7.13	Single Solvent Crystallization	374	
7.14	Mixed Solvent Crystallization	375	

6.3	Sublimation	330
6.3.A	Overview of Sublimation	331
6.3.B	Step-by-Step Procedures	332
	Under Atmospheric Pressure	332
	Under Reduced Pressure (Vacuum Sublimation)	334
	Vacuum Sublimation Summary	337
6.4	Chemical Tests	338
6.4.A	Overview of Chemical Tests	338
6.4.B	Flowcharts	339
6.4.C	Chemical Test Summary	342
6.4.D	Individual Tests	344
	Beilstein Test	344
	Benedict's Test	345
	Bicarbonate Test	347
	Bromine Test	348
	Chromic Acid (Jones) Test	349
	2,4-DNPH (Brady's) Test	350
	Ferric Hydroxamate Test	351
	Iodoform Test	352
	Lucas Test	353
	Permanganate (Baeyer) Test	354
	pH Test	355
	Phenol Test	356
	Silver Nitrate Test	357
	Sodium Iodide (Finkelstein) Test	358
	Tollens Test	359
7.15	Single Extraction	376
7.16	Multiple Extraction	377
7.17	Microscale Extraction	378
7.18	Testing the pH After a Wash	379
7.19	Using Drying Agents	380
7.20	Acid-Base Extraction	381
7.21	Simple Distillation	382
7.22	Fractional Distillation	383
7.23	Vacuum Distillation	384
7.24	Steam Distillation	385
7.25	Rotary Evaporation	386
7.26	Melting Points	387
7.27	Boiling Points (Thiele Tube)	388
7.28	Vacuum Sublimation	389

#### **ABOUT THE AUTHOR LISA NICHOLS**

This resource is the result of an academic sabbatical leave in the 2015-2016 academic year. The goal of the project was twofold: a) to create a free electronic laboratory resource for students, b) to show greater step-by-step detail for organic chemistry lab techniques than is usually shown in print textbooks, so that students could come to lab with greater confidence.

Lisa Nichols (Figure 1) obtained a Bachelor's of Science degree in chemistry from California State University, Chico in 2001 and a Master's degree in organic chemistry from Stanford University in 2003. At the time of the second edition (2017) she had taught chemistry full-time for 14 years at Butte Community College (in Oroville, northern California, near C.S.U. Chico), with an emphasis on teaching majors-level organic chemistry.



**Figure 1:** Author running a chromatography column for this project.

Lisa Nichols would like to express gratitude to the many people who made this project possible and/or enhanced the final product:

- The administration and LTPA committee at Butte College that selected her to receive the sabbatical.
- Special thanks to Dr. David Ball, Professor Emeritus at C.S.U. Chico, who was the primary editor for this textbook. Dr. Ball was always willing to discuss best practices and was also a source of occasional supplies.
- Special thanks to the secondary editors of this textbook: Dr. Carolynn Arpin, Dr. Christopher Nichols (faculty at C.S.U. Chico), Dr. Michael Lodewyk, and Dr. Kristin Milinkevich (faculty at Butte College).
- Dave Carr and Rusty Bogart (staff at Butte College) who were often asked to take pictures when both hands were needed in the photograph. Some of Mr. Carr's best work is shown in Figure 2.
- Occasional photography assistance: Dr. Elizabeth Wannenmacher, Martin Wallace (faculty at Butte College), Dr. Monica So, and Jana Burgess-Henry (faculty at C.S.U. Chico).
- Discussion of physics-related phenomena: Dr. Robert White (faculty at Butte College).
- Advice on publishing and multimedia: Jerry Garcia and Peter Dahl (staff at Butte College).
- Advice on online curricular materials: Matt Evans and Mark Blaser (faculty at Shasta College).
- For help on improving the accessibility of the textbook: Suzanne Wakim (faculty at Butte College).



Figure 2: Some excellent photographs taken by Dave Carr.

### NOTE TO INSTRUCTORS

This resource is a PDF of an organic chemistry laboratory textbook designed for the undergraduate organic chemistry student who has never before taken a college-level organic chemistry course. It can also be used for students engaging in upper division courses or independent research who wish to have a refresher in basic laboratory techniques. This resource was not designed with graduate students in mind, and therefore may not be all inclusive in laboratory techniques experienced at that level.

One of the main goals of this project was to provide a free, yet quality resource for organic chemistry students. With that in mind, the author Lisa Nichols welcomes feedback on typos, errors, or differences in opinion that readers come across. Please send comments to: <u>nicholsli@butte.edu</u>.

It was not practical to show every conceivable iteration of every lab technique, and focus was placed on what was considered to be the most commonly encountered methods. Consideration was made to what materials are likely available in teaching labs. For example, a balloon technique was presented for inert atmospheric work as it was thought that gas manifolds are less common in academic labs.

#### COPYRIGHT ISSUES



Figure 3: Creative Commons logo.

This work is licensed under a <u>Creative Commons</u> Attribution-NonCommercial-NoDerivatives 4.0 International License.

In other words, you are allowed to copy and distribute the material under the following terms:

• Attribution: Please give appropriate credit to the author, for example in the following way:

From: Organic Chemistry Laboratory Techniques, Nichols, 2017

- Non-Commercial: This work may not be used for commercial purposes, or those primarily intended for monetary compensation. An exception is that the author allows for reproduction of this text in print format either at cost, or with small profit margins by academic bookstores.
- No Derivatives: This work is not intended to be remixed, transformed, or built upon. However, the author allows for the following exceptions: a) Photograph screenshots and paragraphs from the text can be used in student or instructor presentations, lectures, and handouts with appropriate credit, b) The final chapter on technique summaries may be removed in print copies.

### NOTE TO STUDENTS

This resource is a PDF of an organic chemistry laboratory textbook designed for the undergraduate organic chemistry student who has never before taken a college-level organic chemistry course. It can also be used for students engaging in upper division courses or independent research who wish to have a refresher in basic laboratory techniques. This resource was not designed with graduate students in mind, and therefore may not be all inclusive in laboratory techniques experienced at that level.

All topics listed in the table of contents of each chapter are active links, and upon selection will take you to the beginning of each topic in the PDF. Some paragraphs in the text also contain links, and they will be underlined and a different color than the text (for example a link may look like this - this link takes you to the cover).

In many PDF readers, there is a method to return to a previous section after a link has sent you elsewhere in the document. In Adobe Reader, this can be accomplished by selecting View < Page Navigation < **Previous View**, while in Preview this can be accomplished by selecting **Go < Back**. This method will work even if you have scrolled rather far from where the link sent you.

Bookmarks

Cover / Preface

Chapter 1: General

Techniques

Chromatography

Chapter 4: Extraction

Chapter 5: Distillation

Chapter 6: Miscellaneous

Chapter 7: Summaries

Chapter 3: Crystallization

▶ □ Chapter 2:

ß

П

Ø

There are several ways to find specific information in this text:

1. Use the links in any Table of Contents (In Brief, In Detail, or at the beginning of each chapter) to jump to chapter headings or the topic of interest.

The chapter headings can also be accessed through the Bookmarks view in Adobe Reader (Figure 4a), or the Table of Contents view in Preview (Figure 4b).

- 2. Skip to the page number of interest (for С example if a page has been assigned  $(\uparrow)$  $( \mathbf{1} )$ 1 / 389 reading). In Adobe Reader, this can be Figure 4: a) Bookmarks icon in Adobe Reader (indicated with an accomplished by entering the page arrow), b) Table of contents icon in Preview (indicated with an arrow), c) Skipping to a page icon in Reader, d) Thumbnail icon in Reader. number in the toolbar that looks like Figure 4c. In Preview this is accomplished by using the Go < Go to Page option.
- 3. Use the **Search function** in the PDF reader to search for keywords. In Adobe Reader, this can be accomplished by using the Edit < Find option, and in Preview a search box is located in the top right toolbar with a magnifying glass icon. The keyword hits can be scrolled through to find the topic of interest. Note that in Preview searching for *water bath* will show separate hits for *water* and *bath*, while using quotation marks to enclose "water bath" will shows hits only when the two words are found together. In Adobe Reader, a search for water bath will only show hits when the two words are found together.
- 4. View the PDF as page thumbnails (in Adobe Reader select the icon in Figure 4d, and in Preview select **View < Thumbnails)** and scan through small images of each page to find the appropriate section.

There are many ways to maximize your experience with a PDF. In many PDF readers, you may highlight text or add notes. Students are encouraged to learn about different options in their individual PDF reader.

Q

Cover / Preface

÷ ~

а

Ð

OChem LabTechniques 2ndEd.pdf

Chapter 1: General Techniques

Chapter 2: Chromatography

Chapter 3: Crystallization

Chapter 4: Extraction

Chapter 5: Distillation

Chapter 7: Summaries

Chapter 6: Miscellaneous

Ô

b

d